

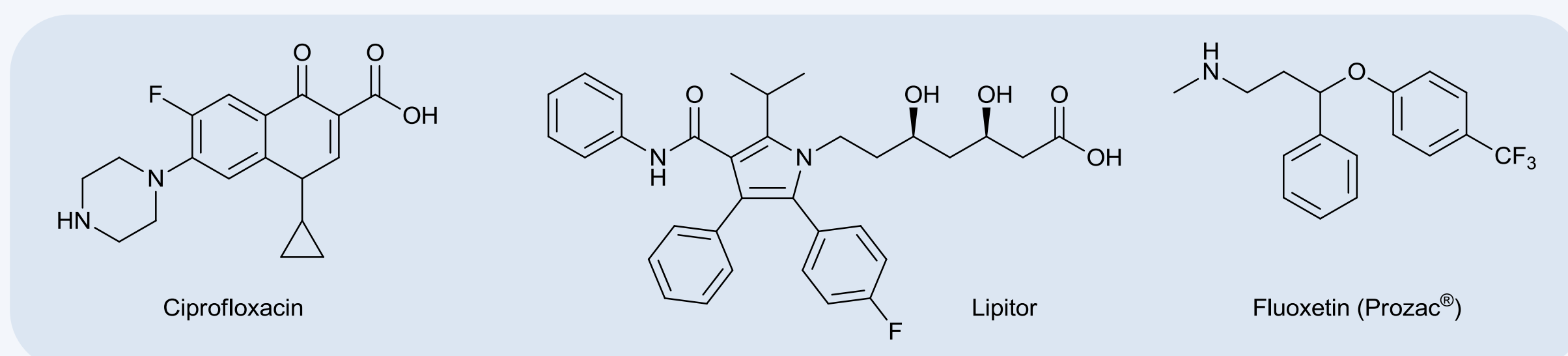
# Synthesis of fluorinated pyrimidinones using fluoro-malonate and fluoro-ketoester « building blocks »

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## Introduction

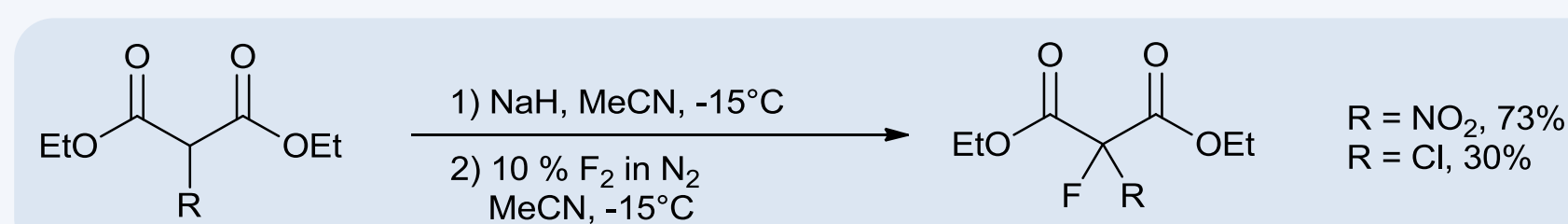
- Significant role of organofluorine chemistry in the majority of the spectacular and technological developments of the past century, due to the specific properties of fluorine atom
- Approximately 30% of new pharmaceutical and agrochemical systems that enter the market bear fluorine atoms



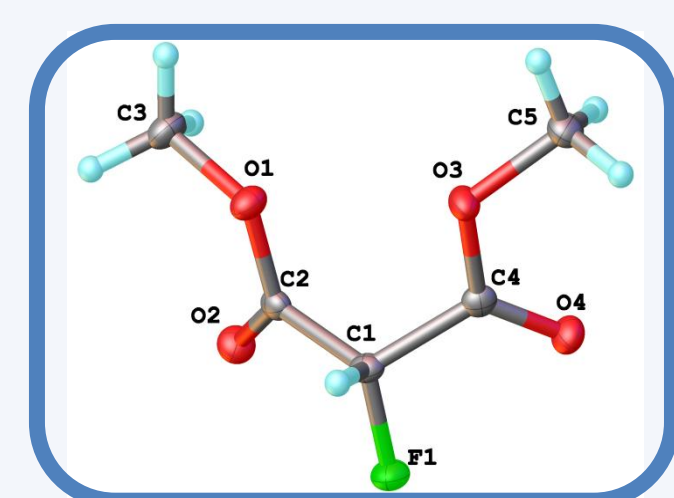
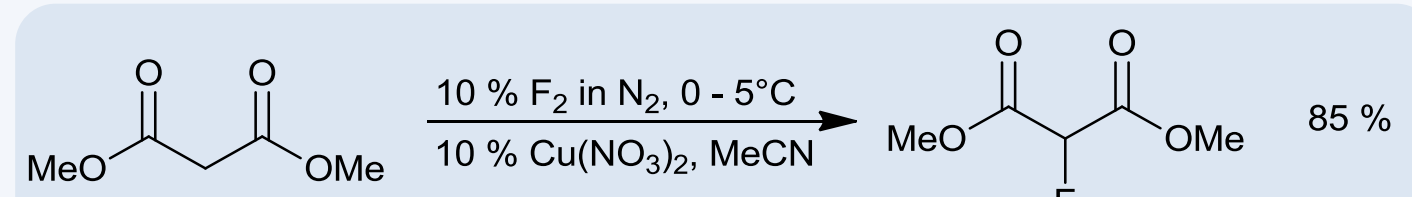
- Development of continuous flow methodology for the selective synthesis of fluoroheterocyclic systems by fluorination of diketones to corresponding fluorodiketones followed by sequential cyclization to appropriate fluoroheterocycles
- Only a few patents dealing with using fluoro-malonates as « building block » for the synthesis of fluorinated biologically active systems

## Synthesis of fluoro-malonates

- Previous studies: replacement of enolic hydrogen atom by fluorine using electrophilic fluorinating agent (NFSI or Selectfluor®) to fluorinate malonate esters
- Direct fluorination of malonates using fluorine diluted in nitrogen
  - Previous studies: reaction of substituted dialkyl malonate salts with elemental fluorine

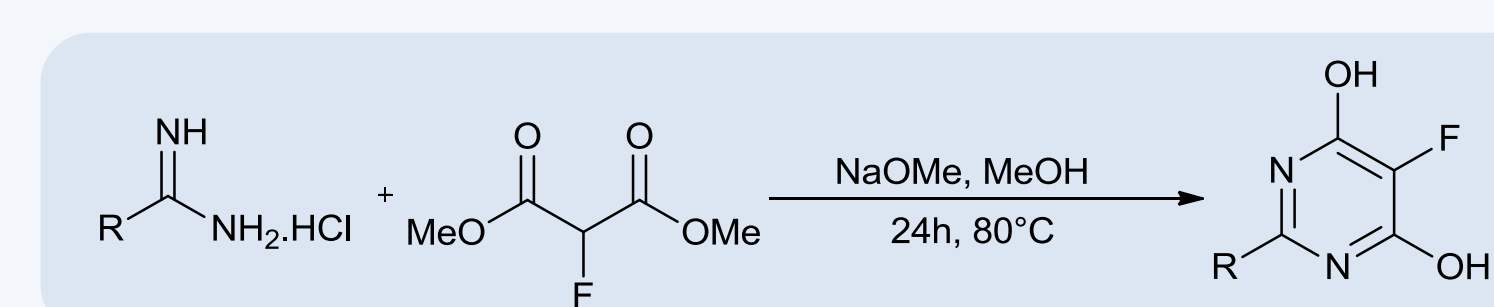


- Malonate substrate activation by adding cat. Cu(NO<sub>3</sub>)<sub>2</sub>



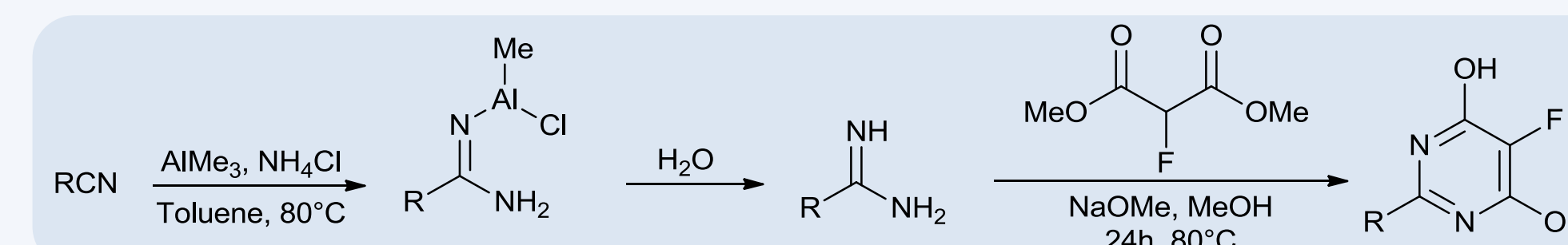
## Synthesis of fluoro-heterocycles

- Synthesis of fluorinated pyrimidinones from previously synthesized monofluoro-malonates
  - with different commercially available amidine hydrochlorides



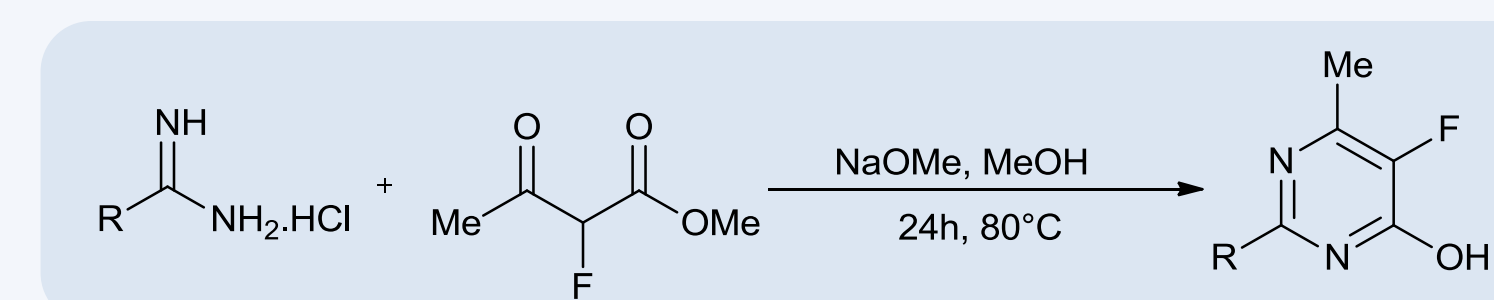
Entry	Substrate (1 eq.)	Yield
(1)	Pyridine-2-carboxamide hydrochloride	72%
(2)	Benzamide hydrochloride	76%
(3)	Cyclopropylcarboxamide hydrochloride	89%
(4)	1 <i>H</i> -Pyrazole-1-carboxamide hydrochloride	51%
(5)	1 <i>H</i> -1,2,4-triazole-1-carboxamide hydrochloride	65%

- with synthesized amidines via Garigipatis procedure

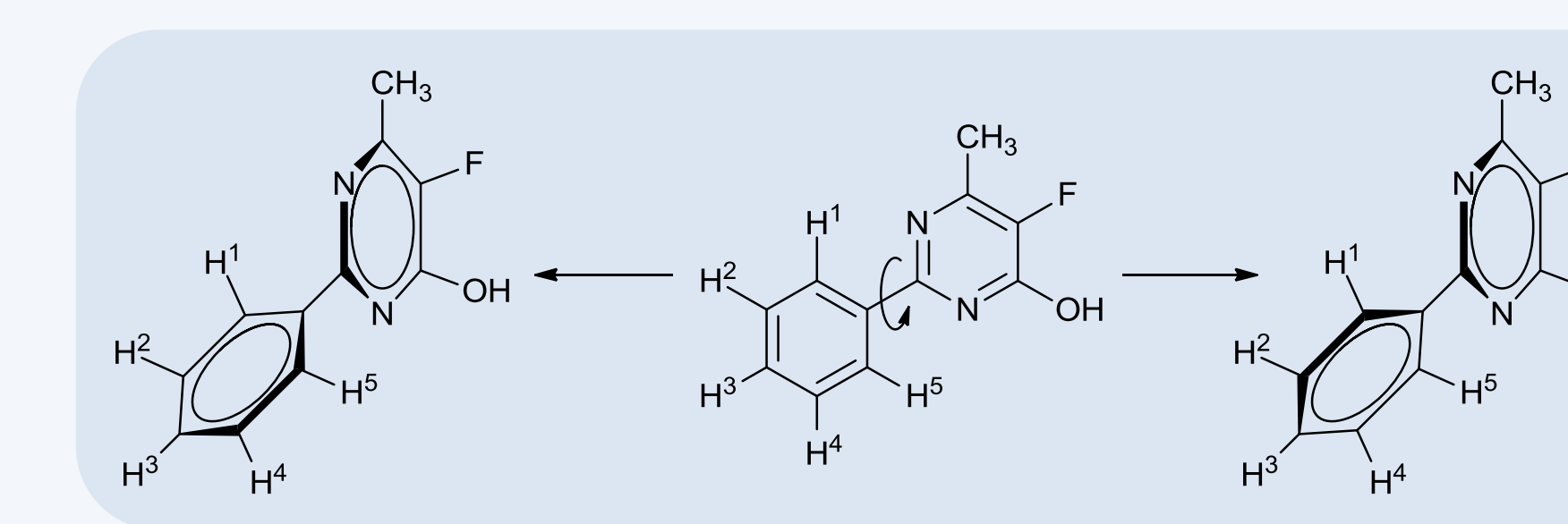
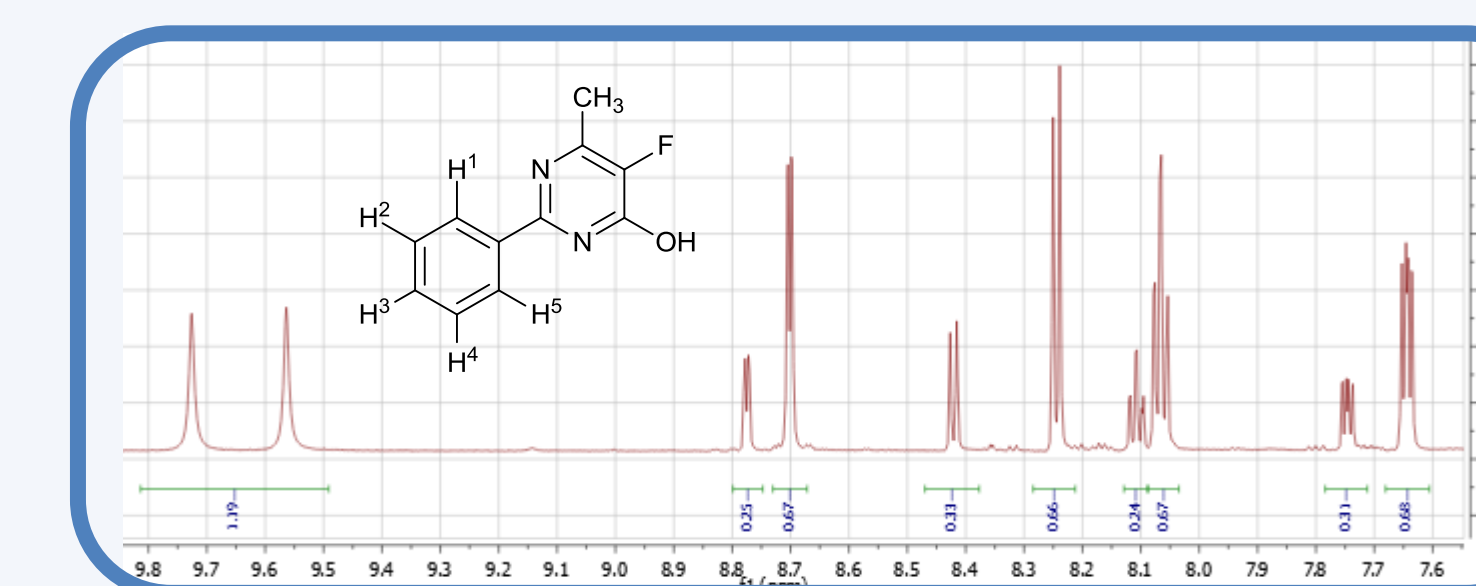
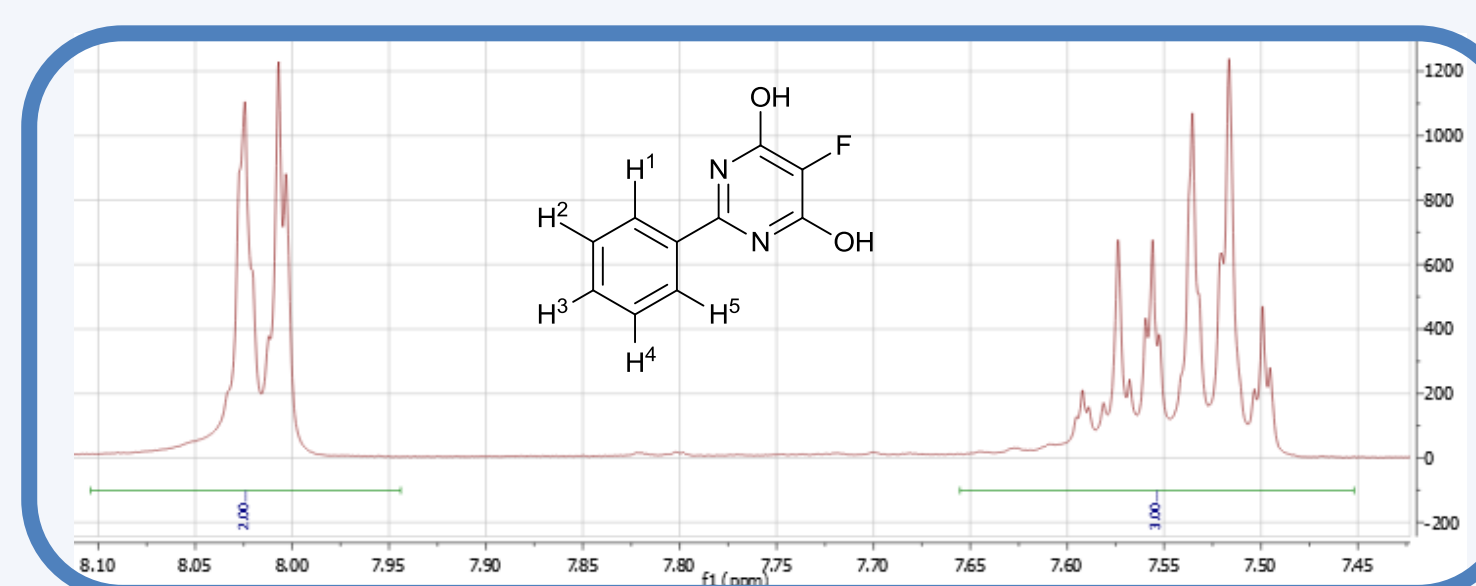


Entry	Substrate (1 eq.)	R	Yield
(1)	Heptyl cyanide	H <sub>3</sub> C-(CH <sub>2</sub> ) <sub>6</sub>	quant.
(2)	Octyl cyanide	H <sub>3</sub> C-(CH <sub>2</sub> ) <sub>7</sub>	quant.
(3)	Undecyl cyanide	H <sub>3</sub> C-(CH <sub>2</sub> ) <sub>10</sub>	80%

- Structural comparison between products from monofluoro-ketoester and monofluoro-malonate
  - Application of the same synthetic method to monofluoro-ketoester substrates
  - Synthesis with two different commercially available amidine hydrochlorides
  - Synthesis of rotamers confirmed by <sup>1</sup>H, <sup>13</sup>C, <sup>19</sup>F NMR spectroscopy and GC-MS spectrometry



Entry	Substrate (1 eq.)	Yield
(1)	Pyridine-2-carboxamide hydrochloride	97%
(2)	Benzamide hydrochloride	quant.



## Conclusions

- Large scale synthesis of fluoro-malonate esters using direct fluorination reactions (10% F<sub>2</sub> in N<sub>2</sub>)
- Synthesis of a range of fluoro-pyrimidinones using either fluoro-malonates or fluoro-ketoesters as « building blocks »
- In case of using fluoro-ketoesters: proof of the synthesis of rotamers by structural analysis

## References

- Chambers R. D., Hutchinson J., *J. Fluorine Chem.* **1998**, *92*, 45-52
- Romedor G., *e-EROS Encycl. of Reagents in Organic Synthesis: Diethyl Malonate*. John Wiley & Sons, **2001**
- Corey E. J., Cheng X.-M., *The logic of chemical synthesis*, Wiley-Interscience, New York, **1995**
- Warren S., Wyatt P., *Org. Synth.: Strategy and Control* (2nd ed.), John Wiley and Sons, Chichester, **2007**
- Warren S., Wyatt P., *Org. Synth.: The Disconnection Approach* (2nd ed.) John Wiley and Sons, Chichester, **2008**
- Harsanyi A., Sandford G., *Org. Process Res. Dev.* **2014**, *18*, 981-992
- Miller T. A., Sloman D. L., Stanton M. G., Wilson K. J., Witter D. J., WO2007087129. **2007**

- Chambers R. D., Hutchinson J., Thomson J., *J. Fluorine Chem.* **1996**, *78*, 165-166
- Warren S., *Designing Org. Synth.: The Synthron Approach*, John Wiley and Sons, New York, **1978**
- Garigipati R. S., *Tetrahedron Letters* **1990**, *31*, 14, 1969-1972
- Ibrahim M. K., Kagaruki S. R. F., *Bull. Chem. Soc. Japan* **1985**, *11*, 2169-2176
- Moss R. A., Ma W., Merrer D. C., Xue S., *Tetrahedron Letters* **1995**, *36*, 48, 8761-8764
- Reddy D. S., Shibata N., Nagai J., Nakamura S., Toru T., Kanemasa S., *Angew. Chem. Int. Ed.* **2008**, *47*, 164-168
- Close J., Heidebrecht R. W., Kattar S., Miller T. A., Sloman D., Stanton M. G., Tempest P., Witter D. J., WO2007055941. **2007**

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